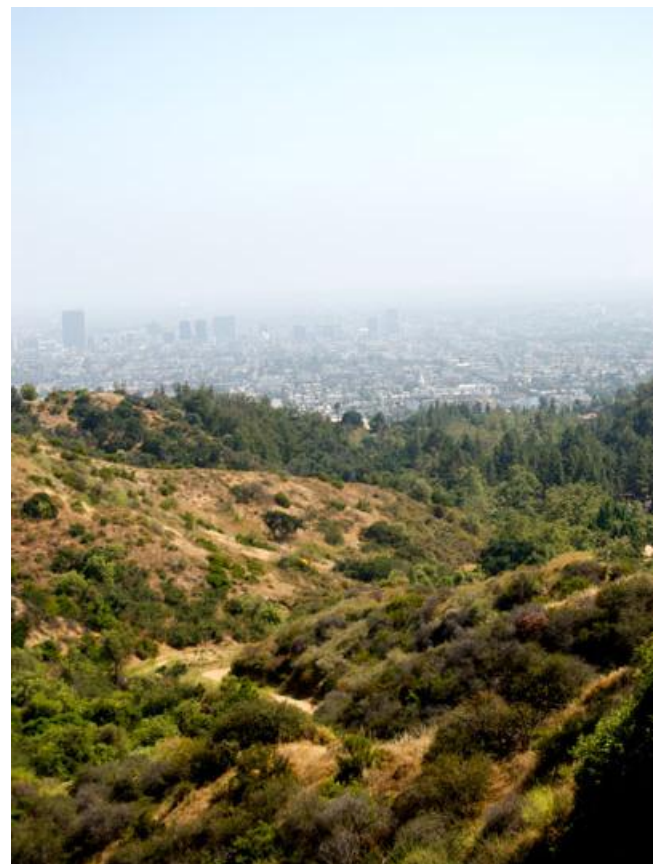


A Tale of Two Extremes: Contrasting NH_3 at the Bakersfield and Pasadena Supersites



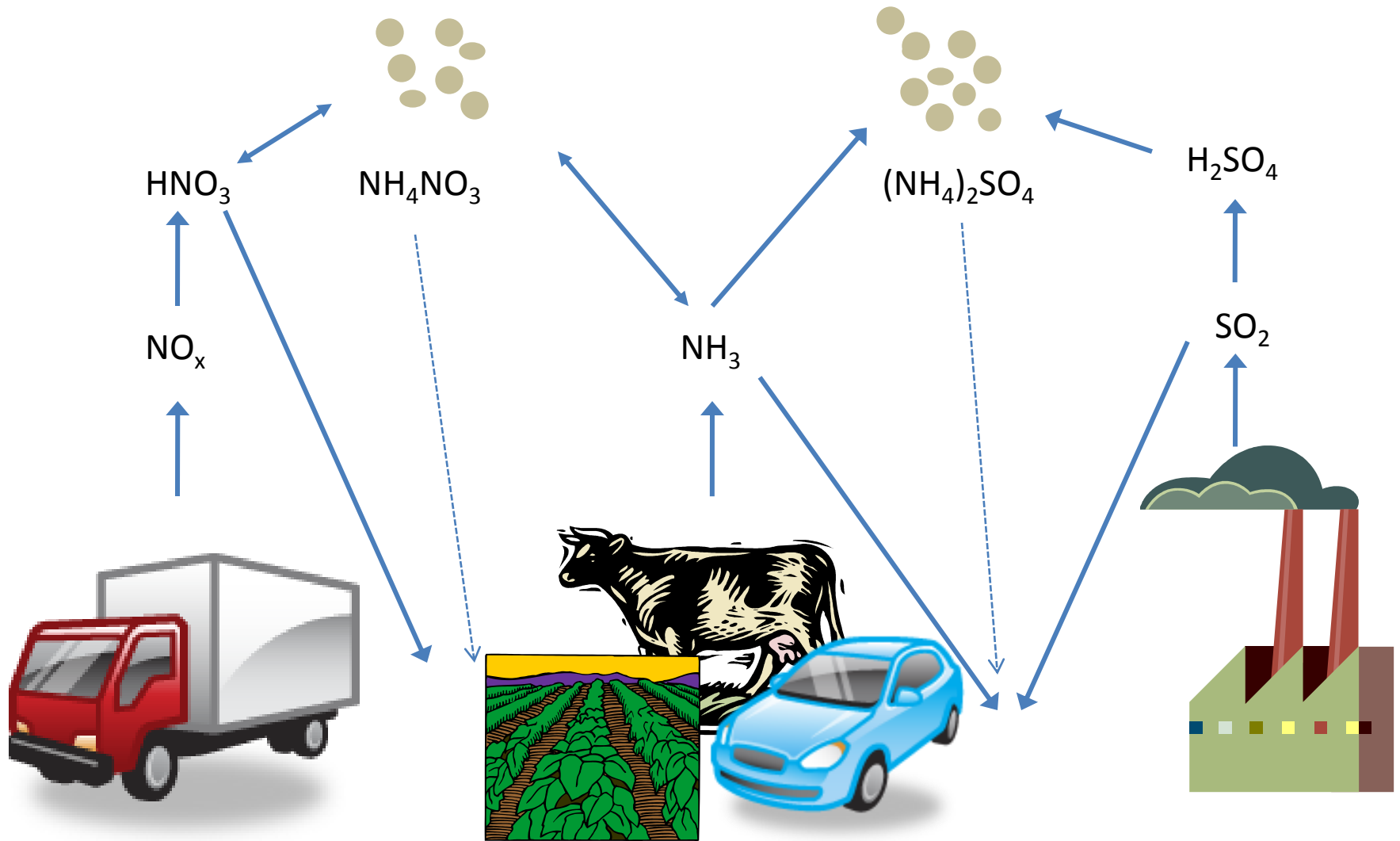
Jennifer Murphy
Milos Markovic
Trevor VandenBoer
Raluca Ellis

Department of Chemistry
University of Toronto

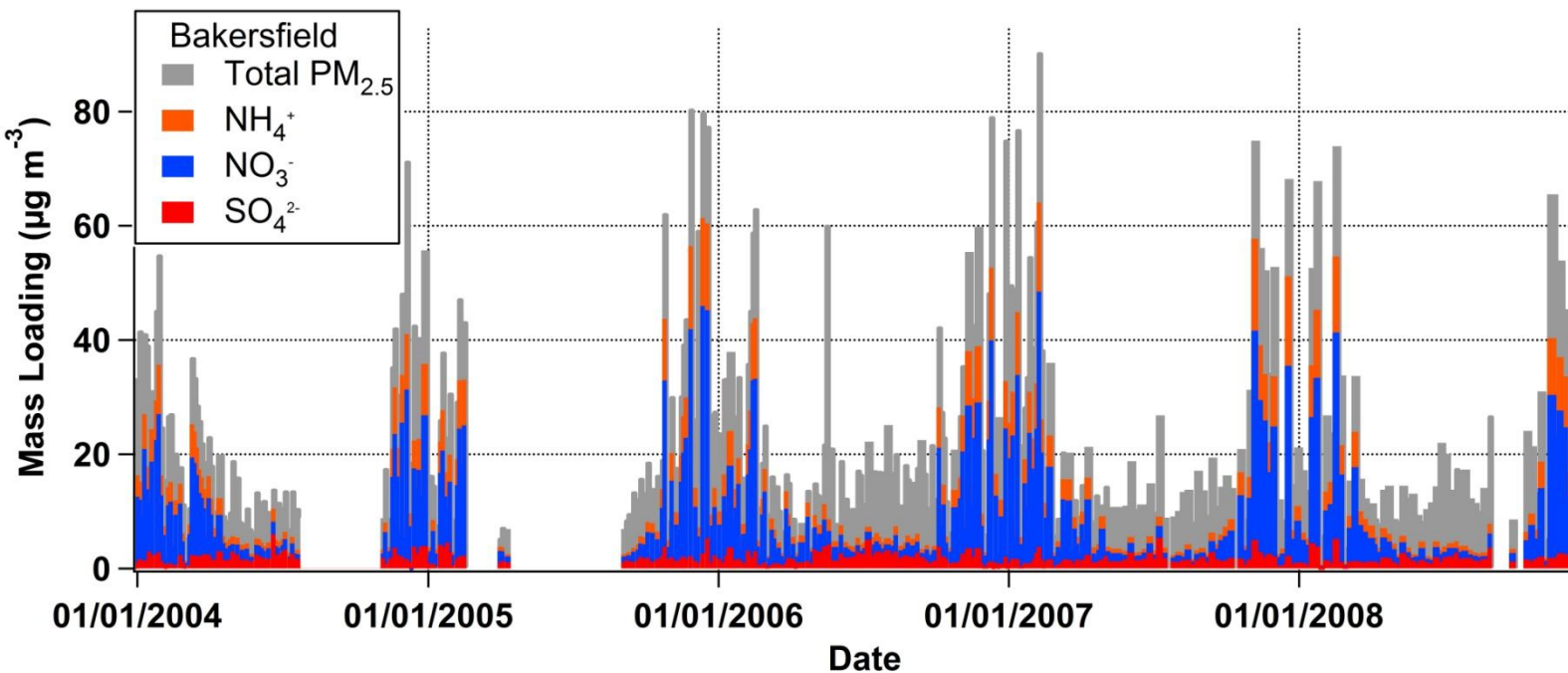


Ammonia and Particle Formation

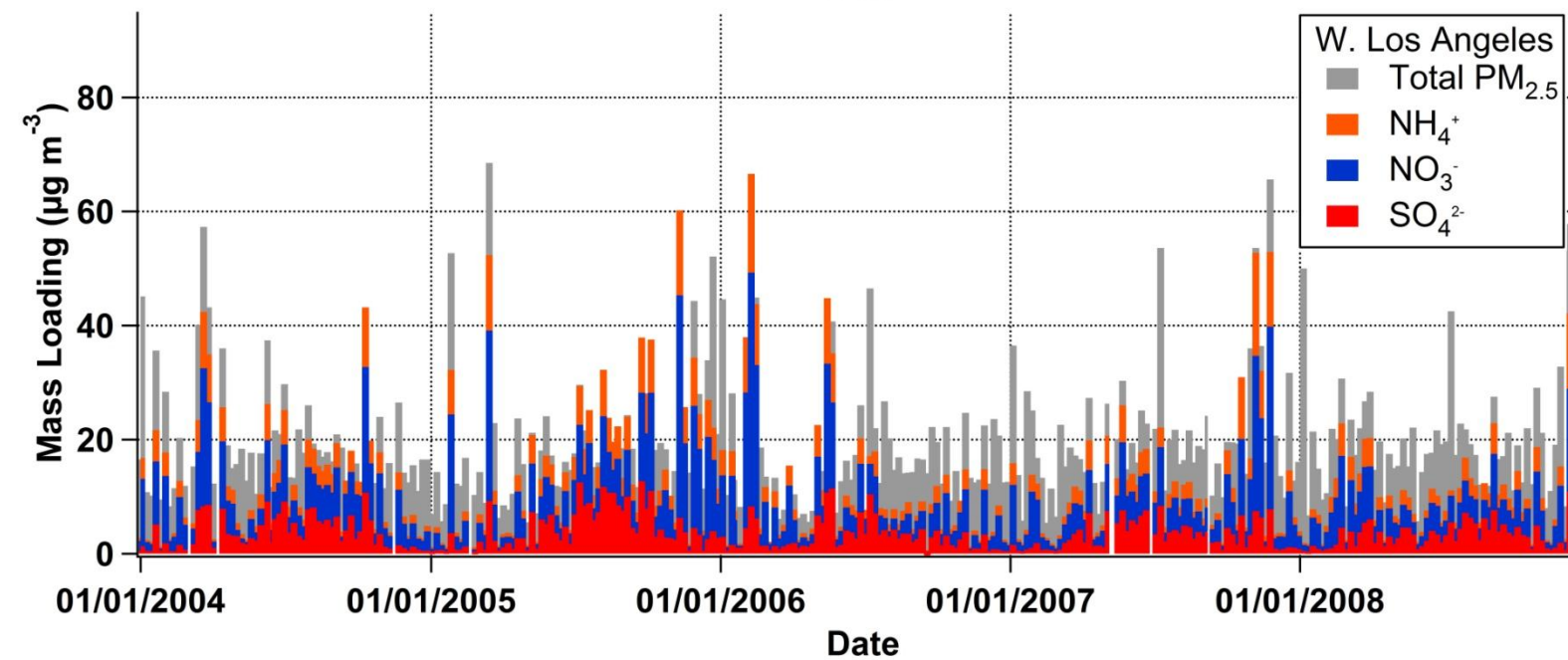
Highly coupled and non-linear



Historical CARB data

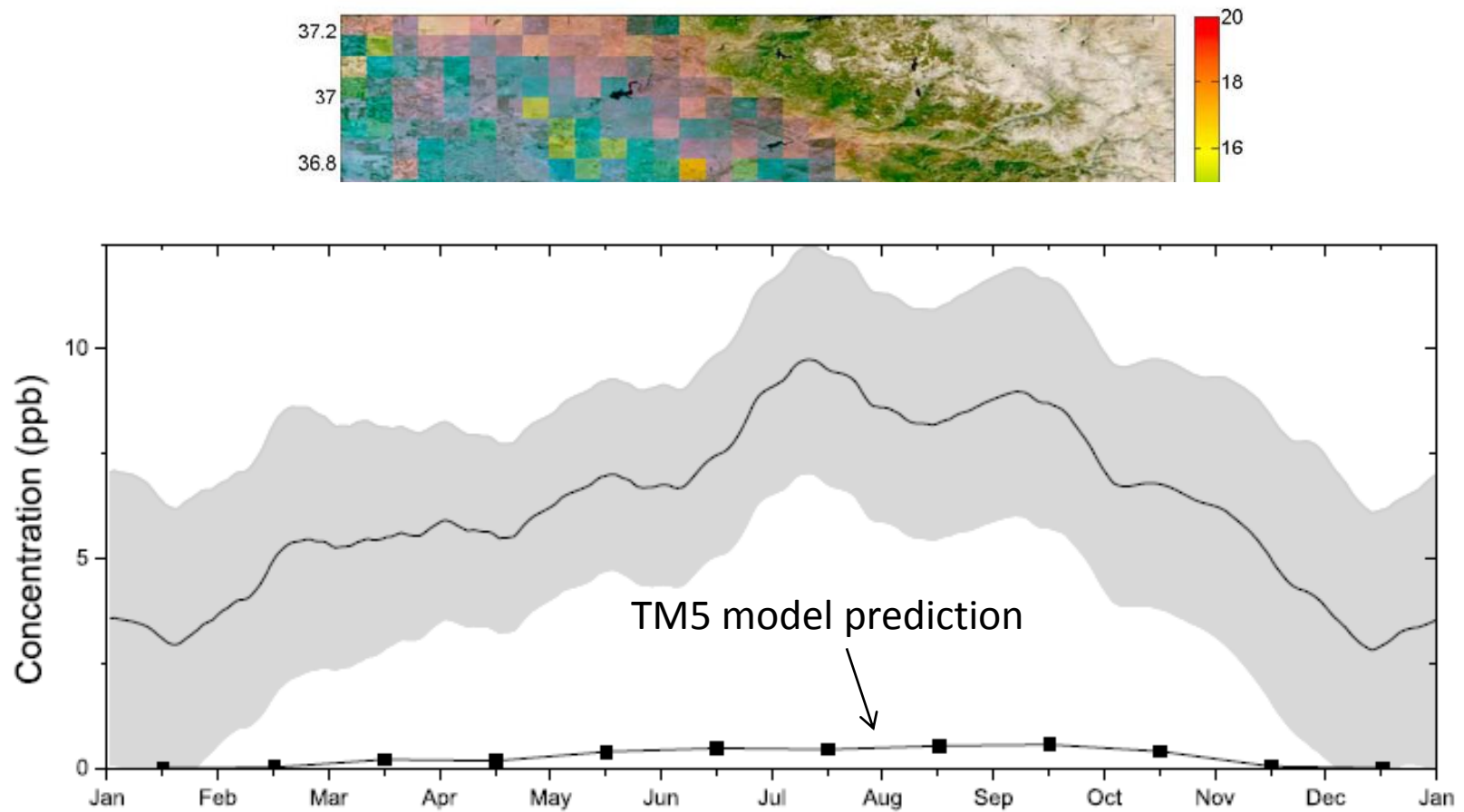


seasonality in
 NH_4NO_3



seasonality in
 $(\text{NH}_4)_2\text{SO}_4$

NH₃ in the San Joaquin Valley



(a) NH₃ concentration at 700m (IASI morning orbit)

Mobile Emissions of NH_3

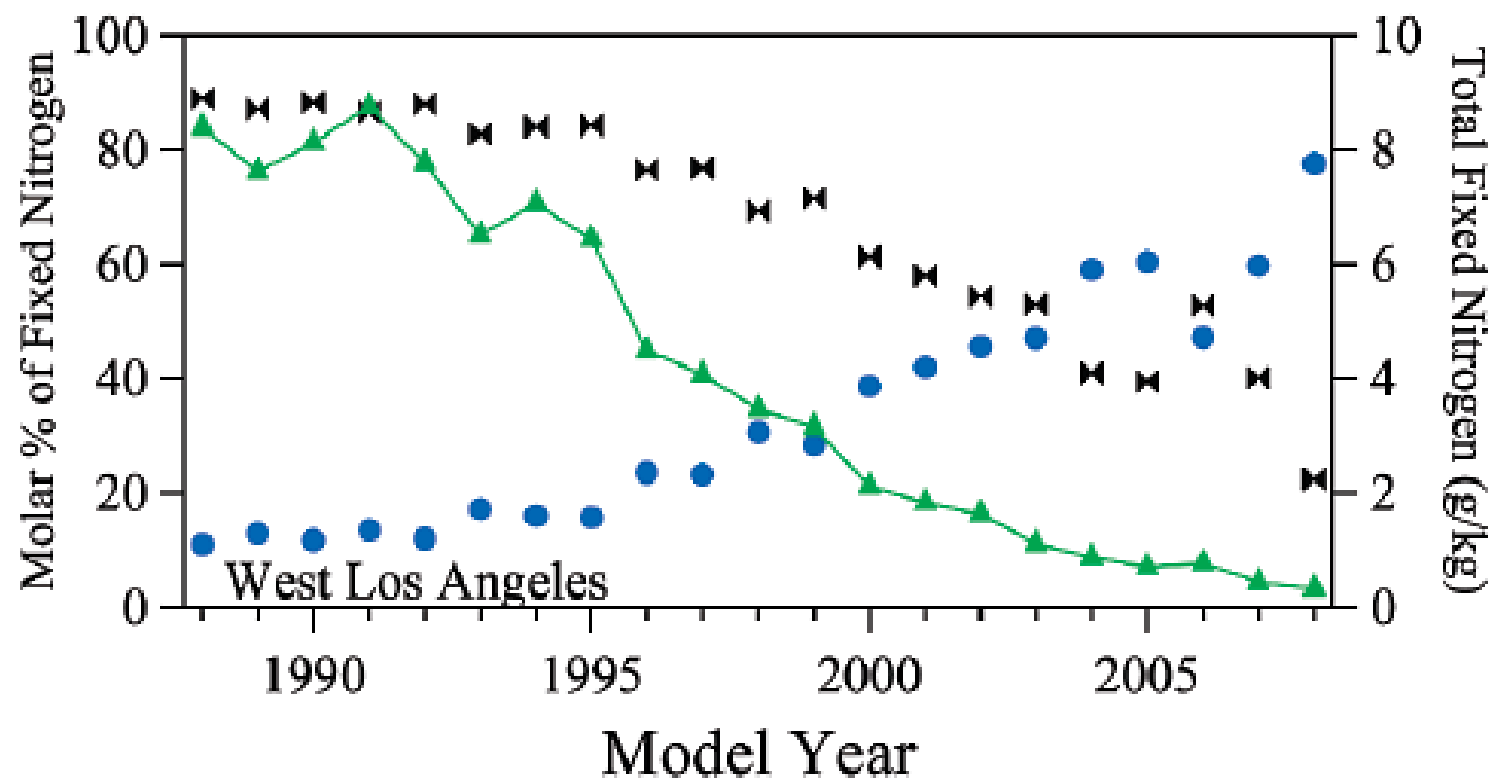
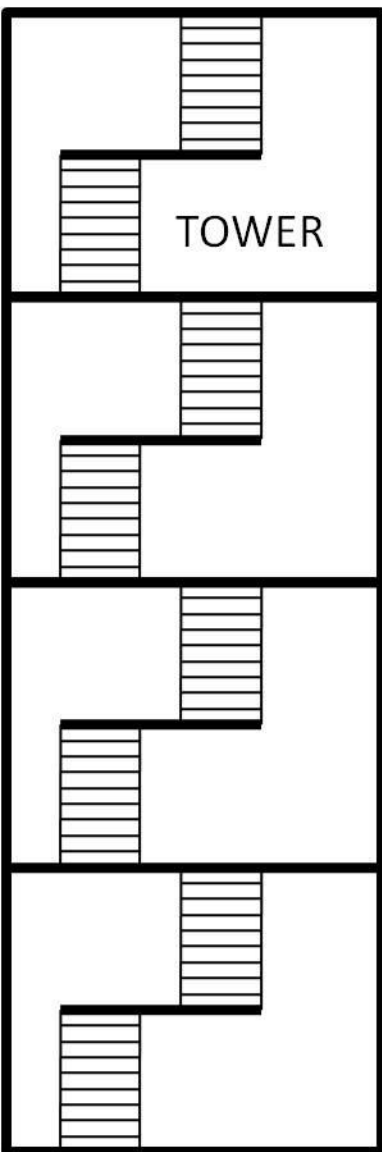


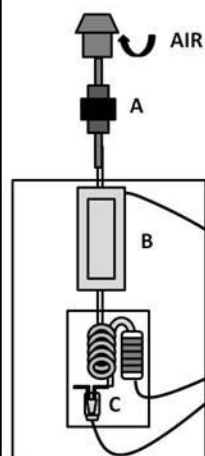
FIGURE 5. Total fixed nitrogen in g/kg (line and triangles, right axis) with the molar percent composition distributed between the NO_x (bowties, left axis) component and the NH_3 component (circles, left axis).

Mobile emissions of NH_3 comparable to NO_x in new, aggressively driven, vehicles

AIM-IC in Bakersfield



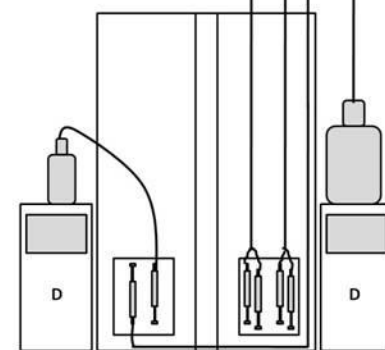
- A – PM_{2.5} impactor
- B – membrane denuder
- C – particle collection chamber
- D – ion chromatographs



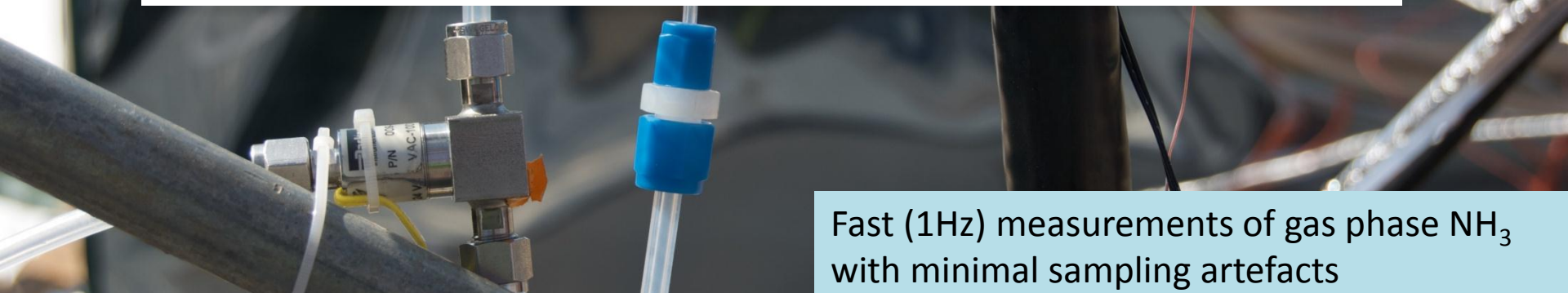
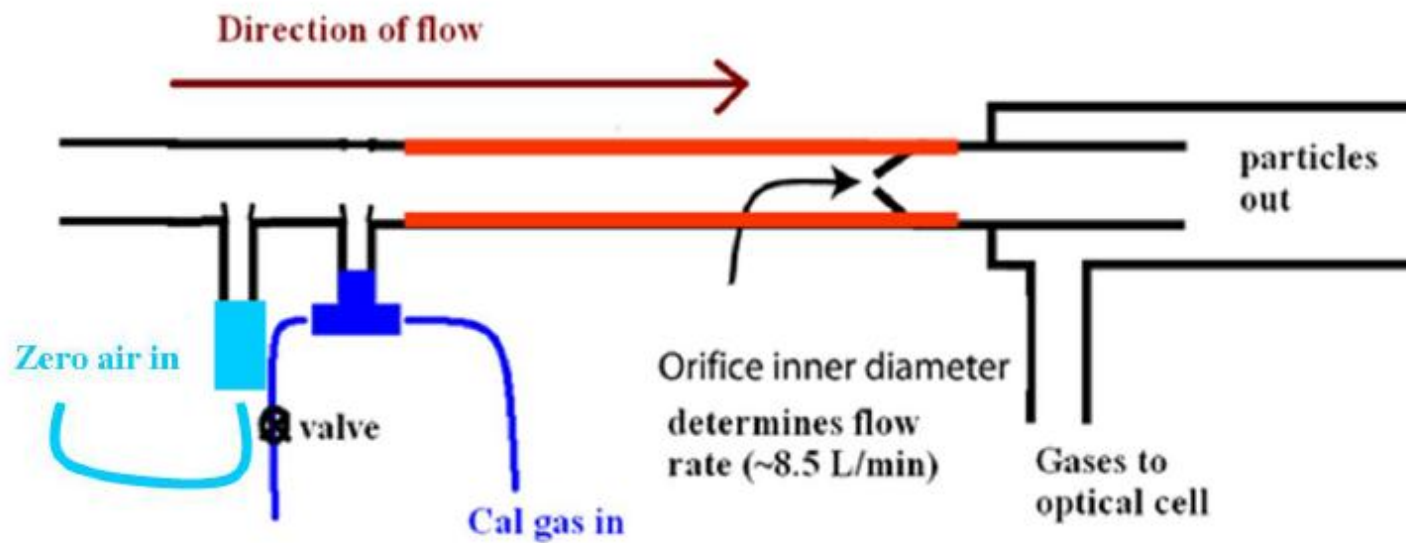
~ 20 m

TRAILER

Hourly gas and PM_{2.5} measurements of
 SO_4^{2-} , NO_3^- , Cl^- , Br^- , NO_2^- , organic acids
 NH_4^+ , K^+ , Na^+ , Mg^{2+} , Ca^{2+} , amines

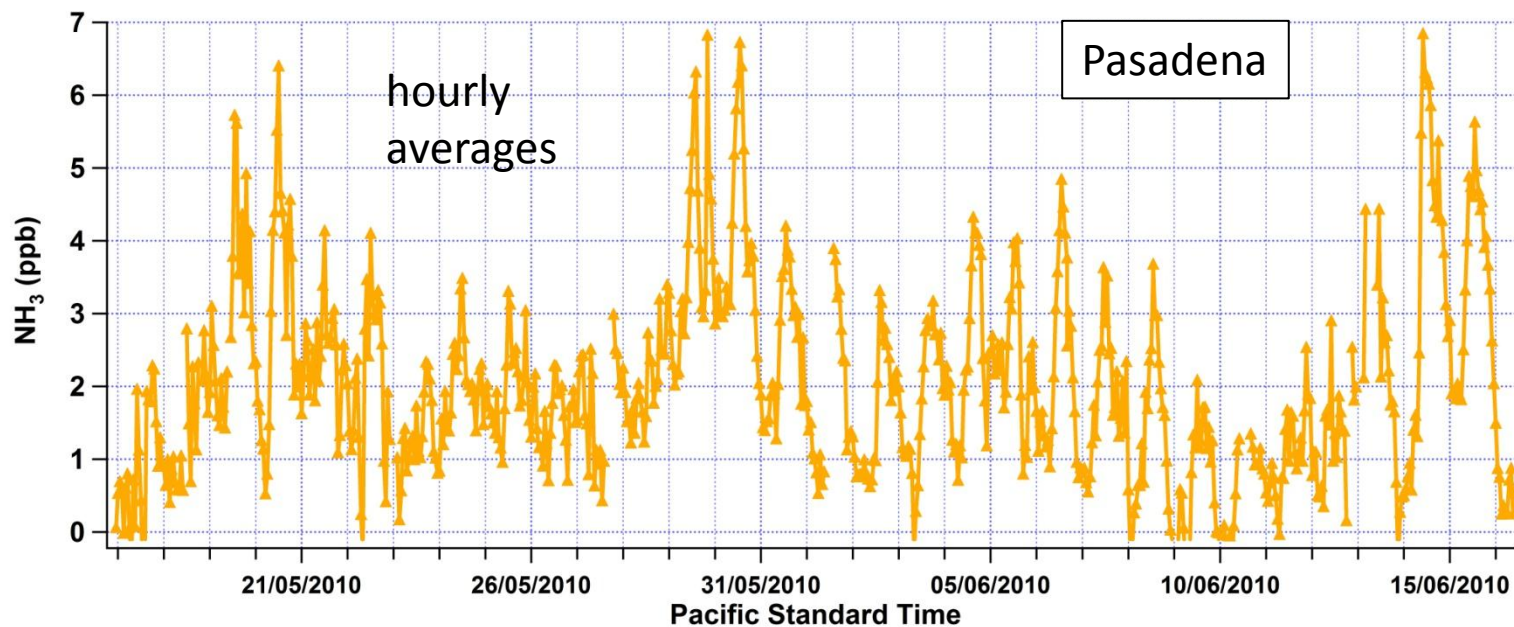
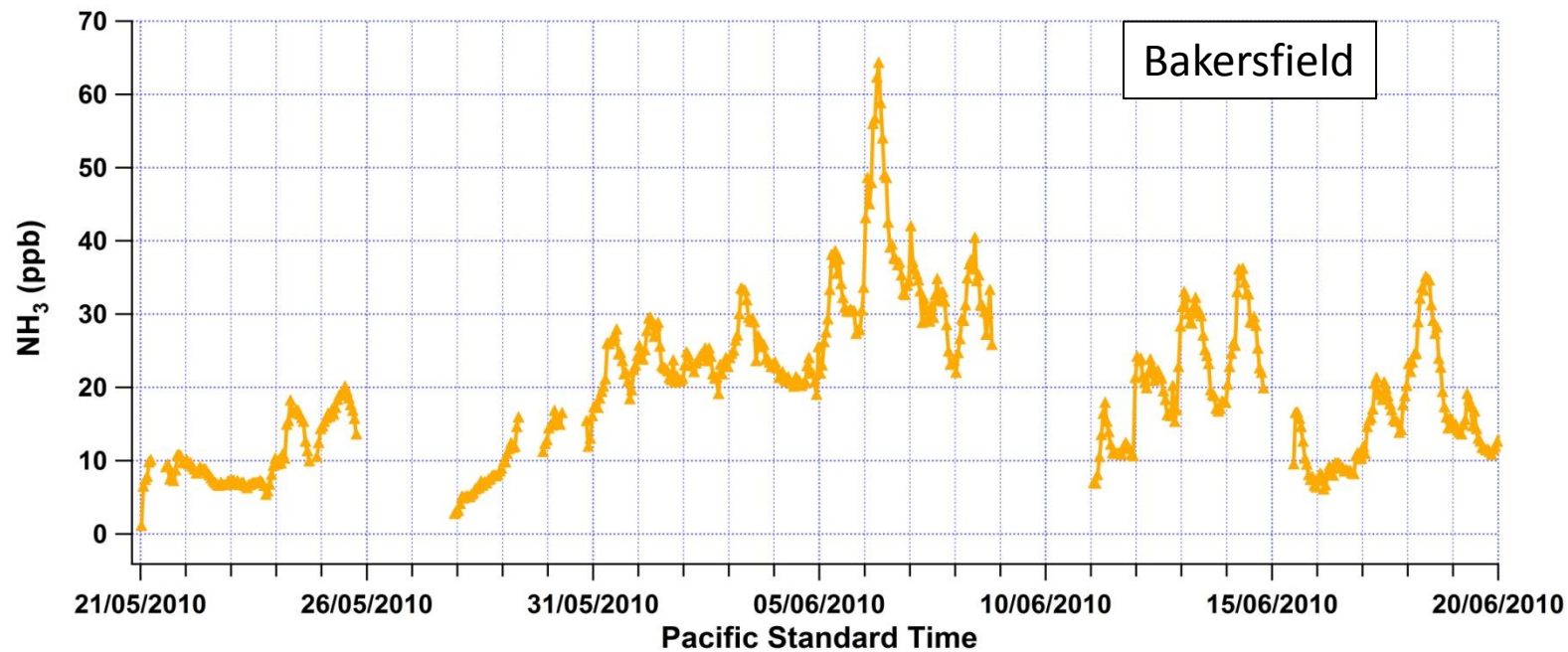


QC-TILDAS in Pasadena



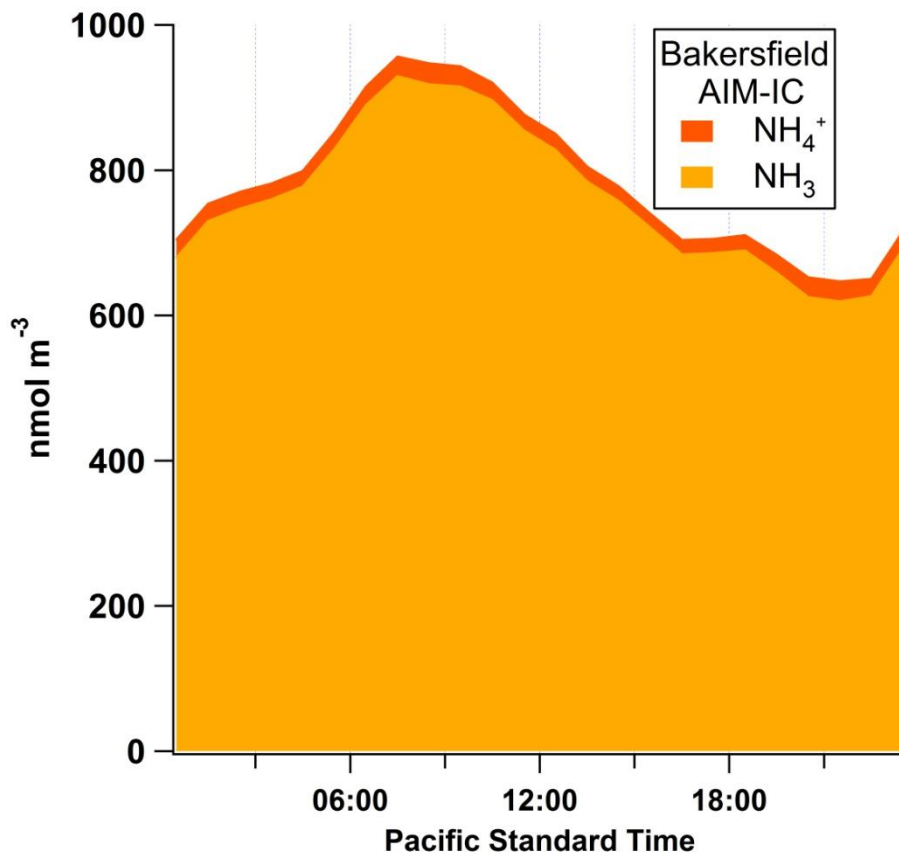
Fast (1Hz) measurements of gas phase NH_3 with minimal sampling artefacts

Ammonia Time Series

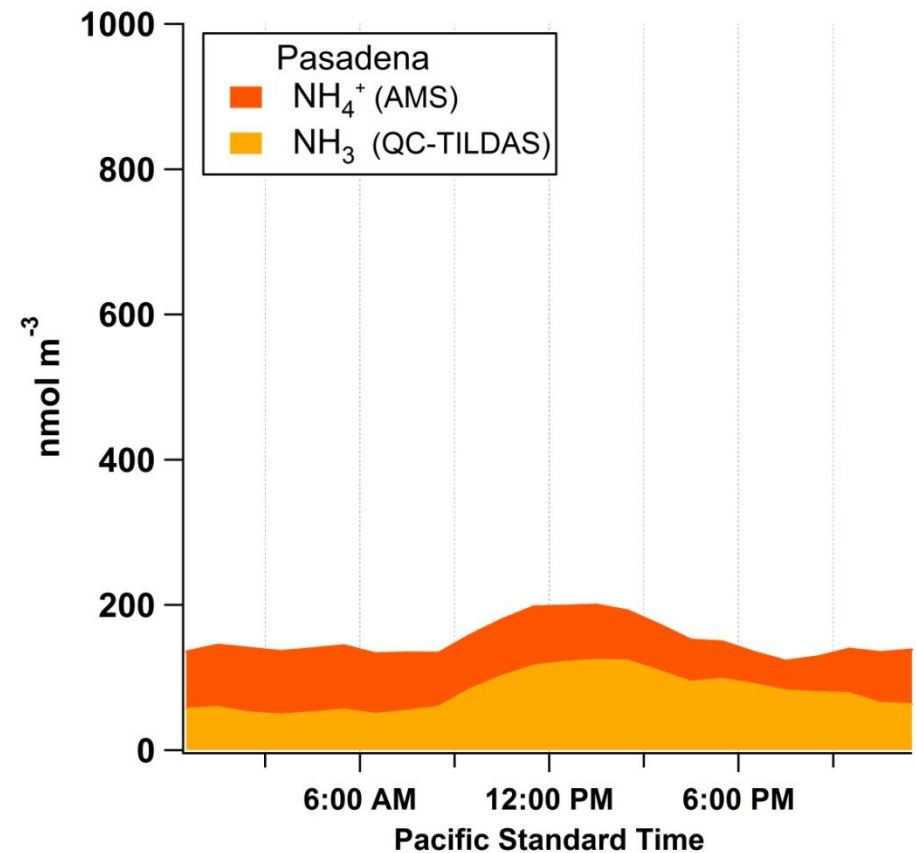


Diurnal Profiles of NH_x Partitioning

Ammonia is controlled by emissions and dilution

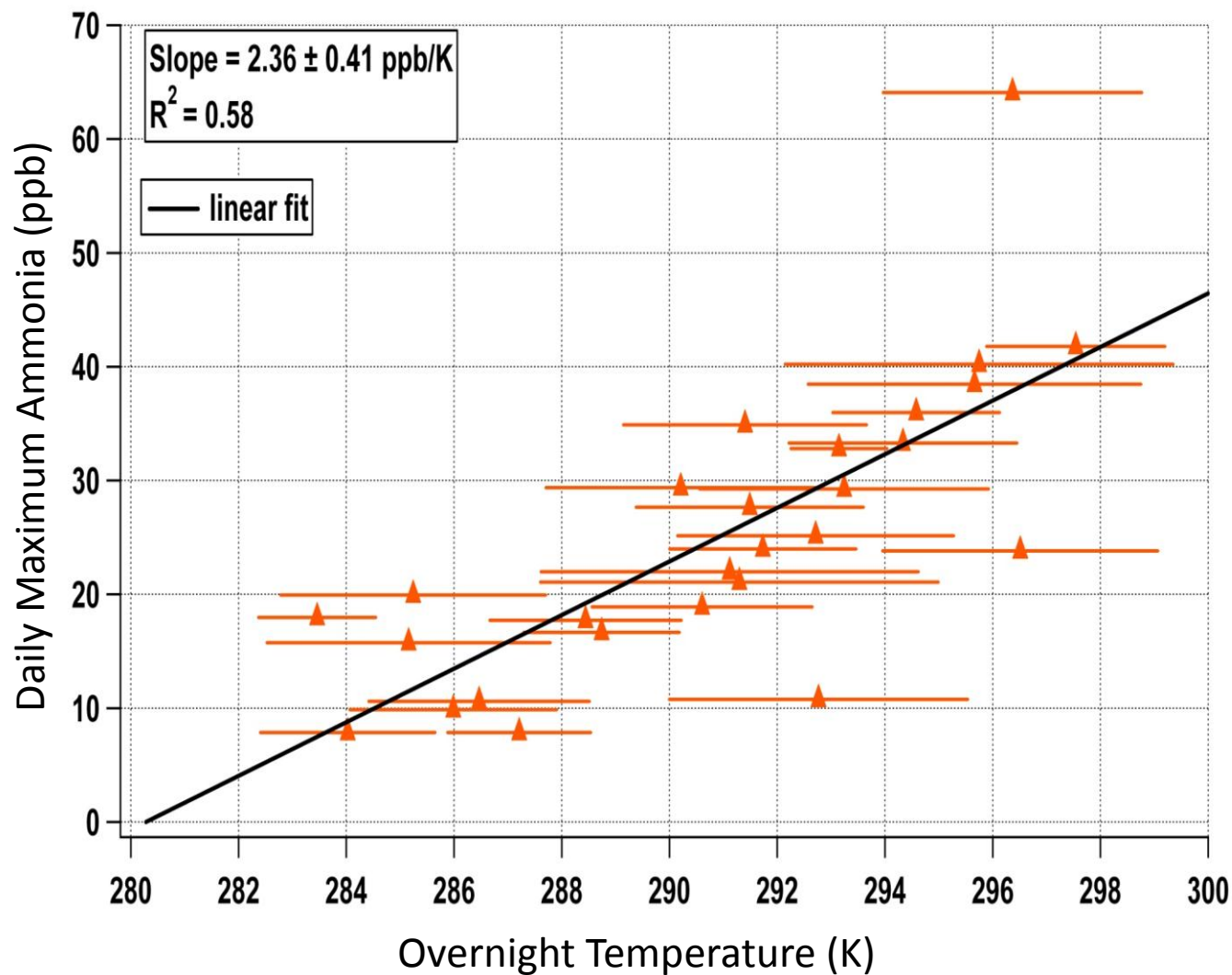


Ammonia is controlled by partitioning and transport

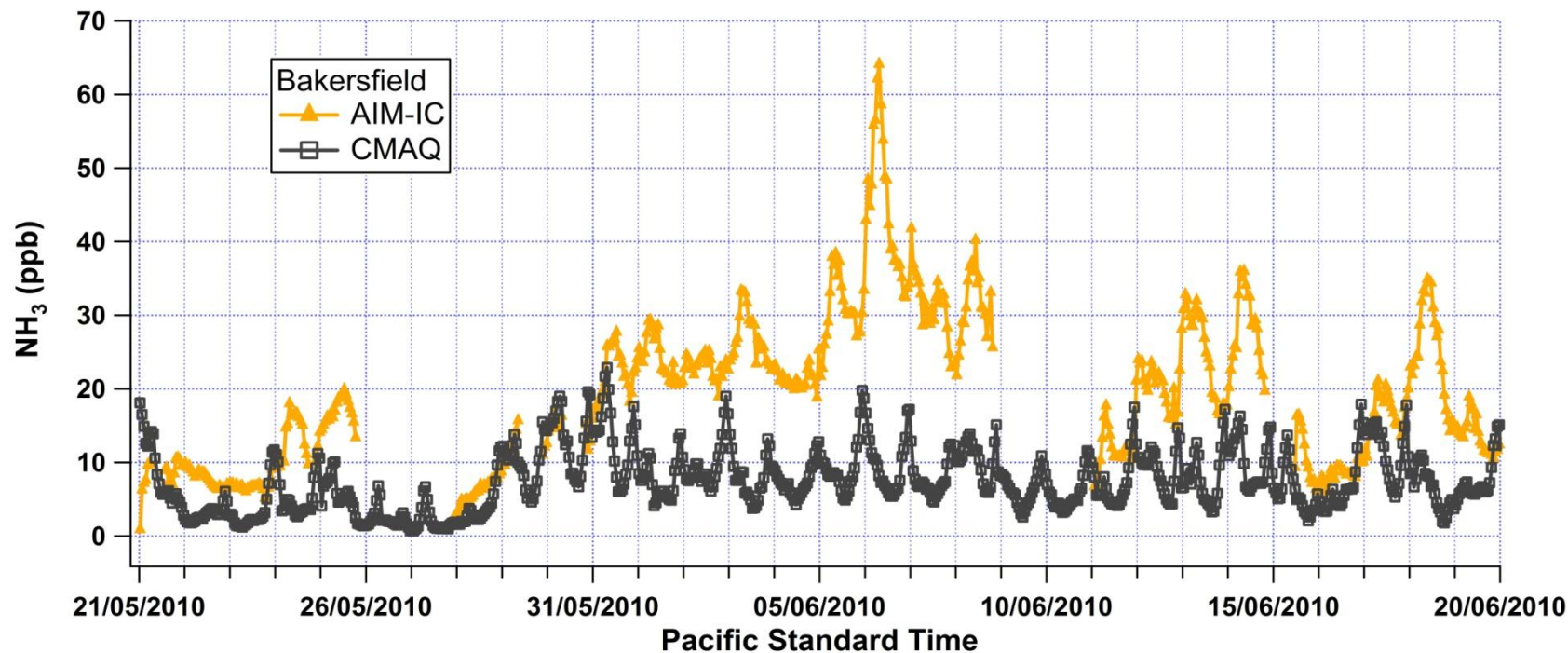


AMS data from Jimenez group, CU

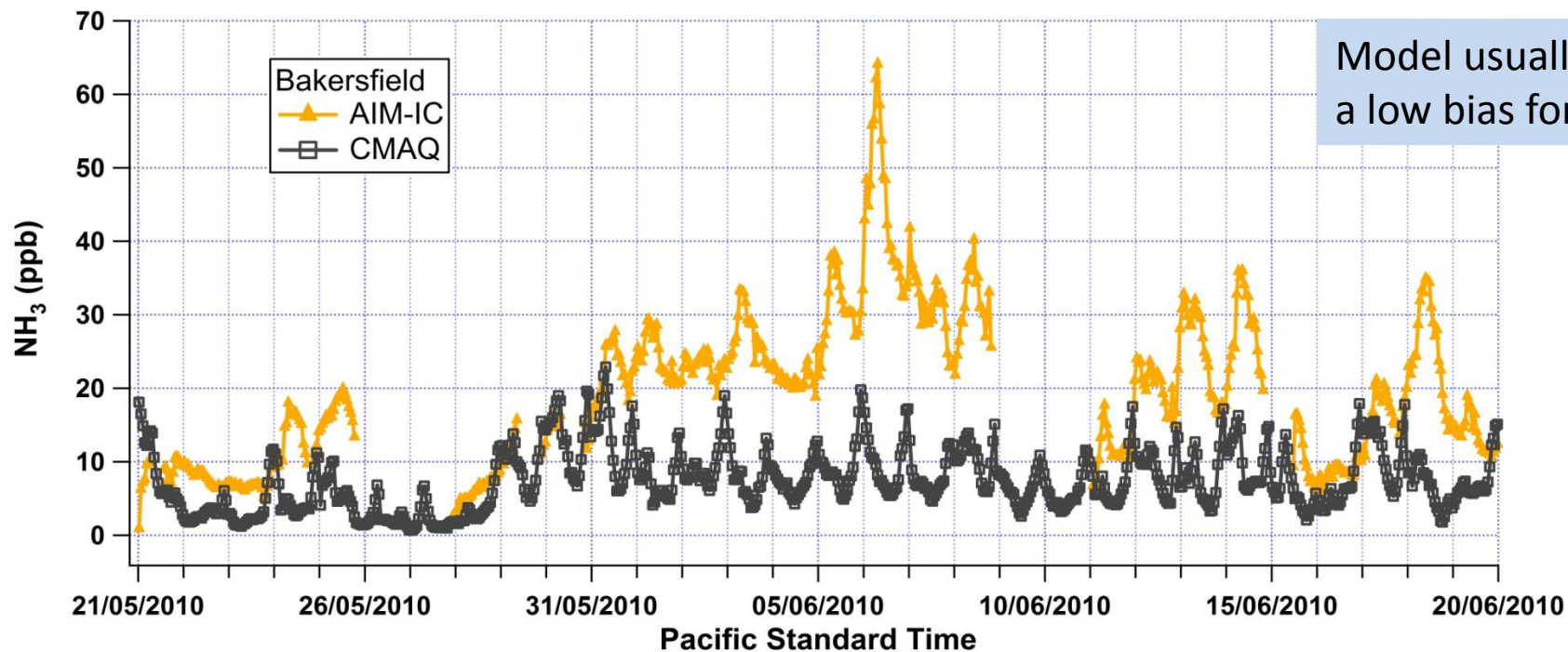
Temperature and NH₃ at Bakersfield



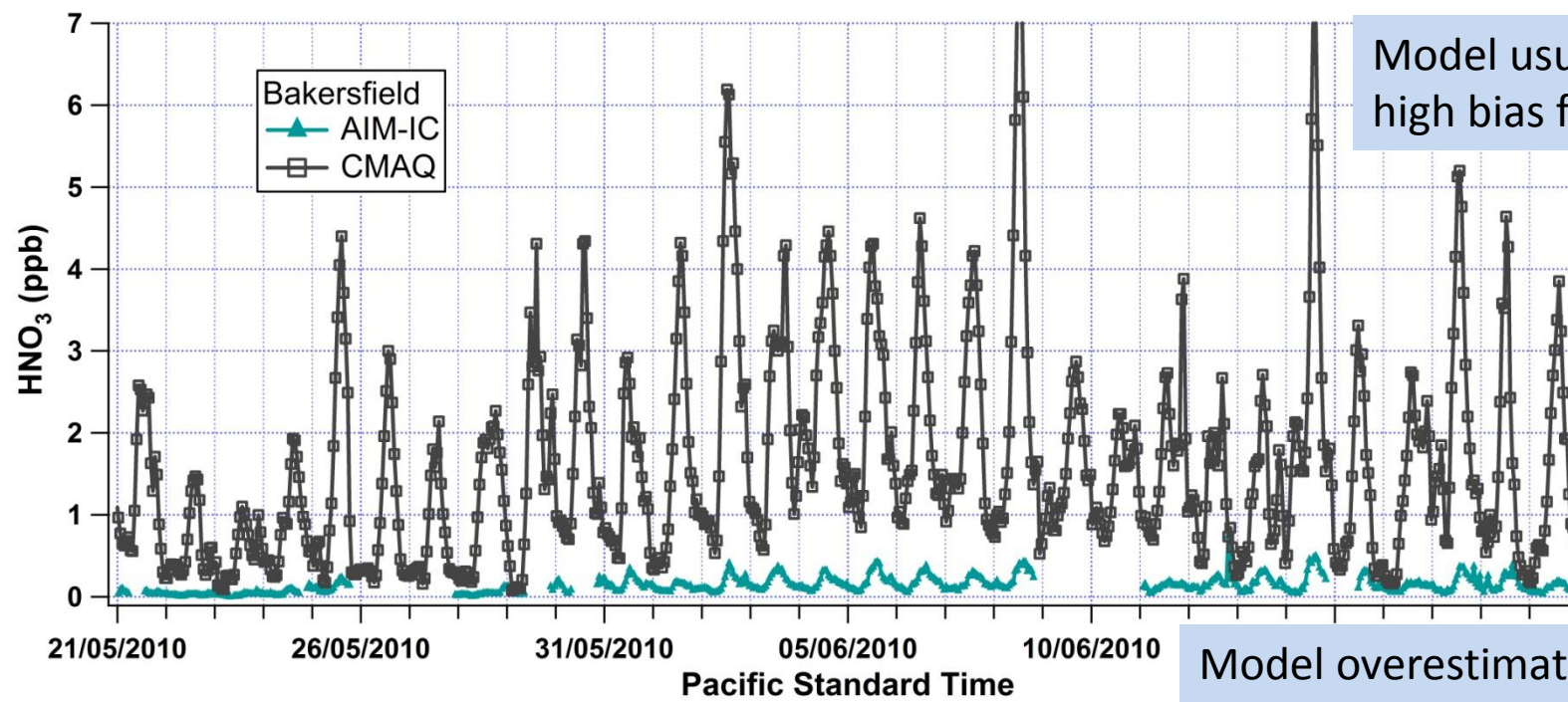
Comparison with Model



CMAQ data courtesy of Kirk Baker, US EPA



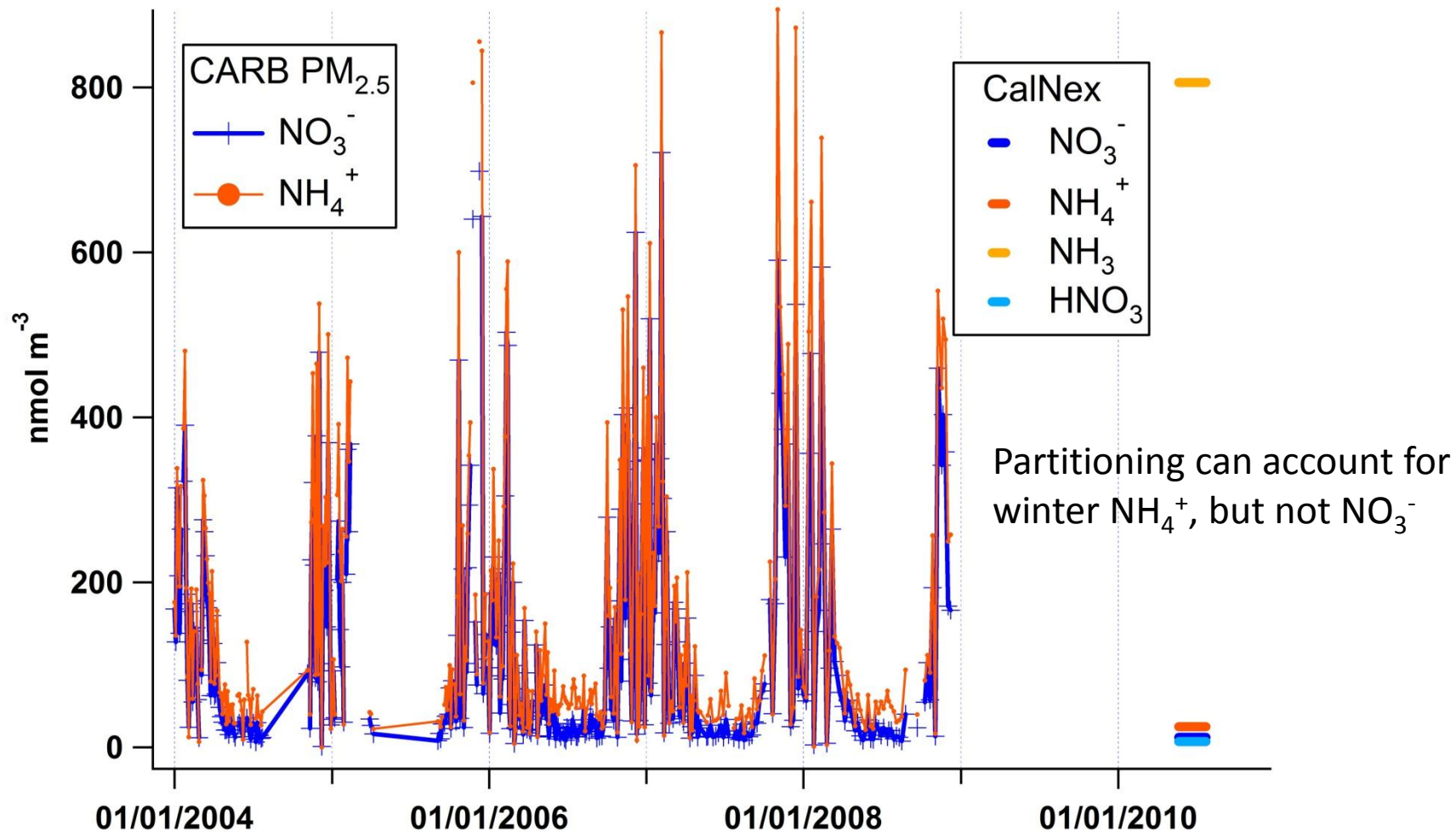
Model usually has
a low bias for NH₃



Model usually has
a high bias for HNO₃

Model overestimates NH₄NO₃

Historical Record at Bakersfield



Summary

- NH_x is ~4 times larger at Bakersfield than at Pasadena
- different processes control NH_3 at the two ground sites
- CMAQ does not represent Bakersfield NH_3 well
(emissions or partitioning)

Additional Questions

- aerosol pH
- influence of/on organic acids
- relative contributions of NH_3 emissions sources

July 29, 2005

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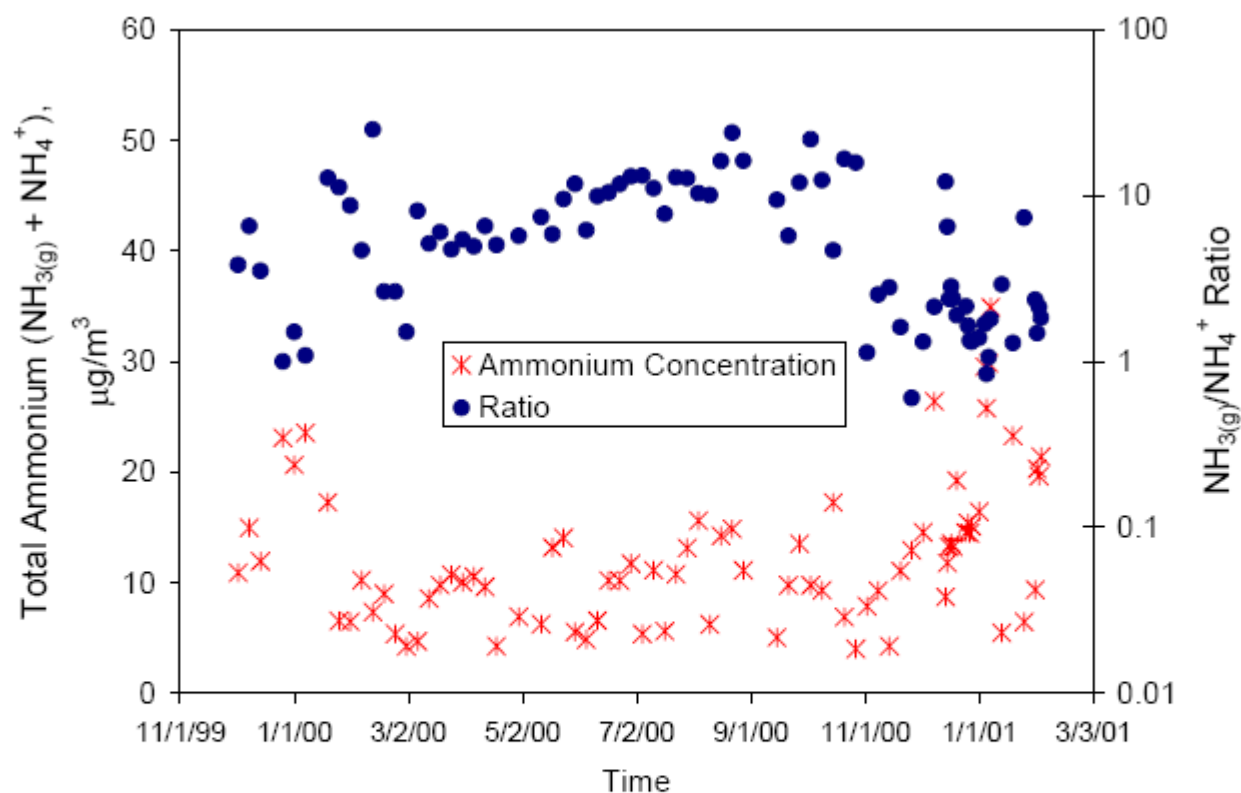
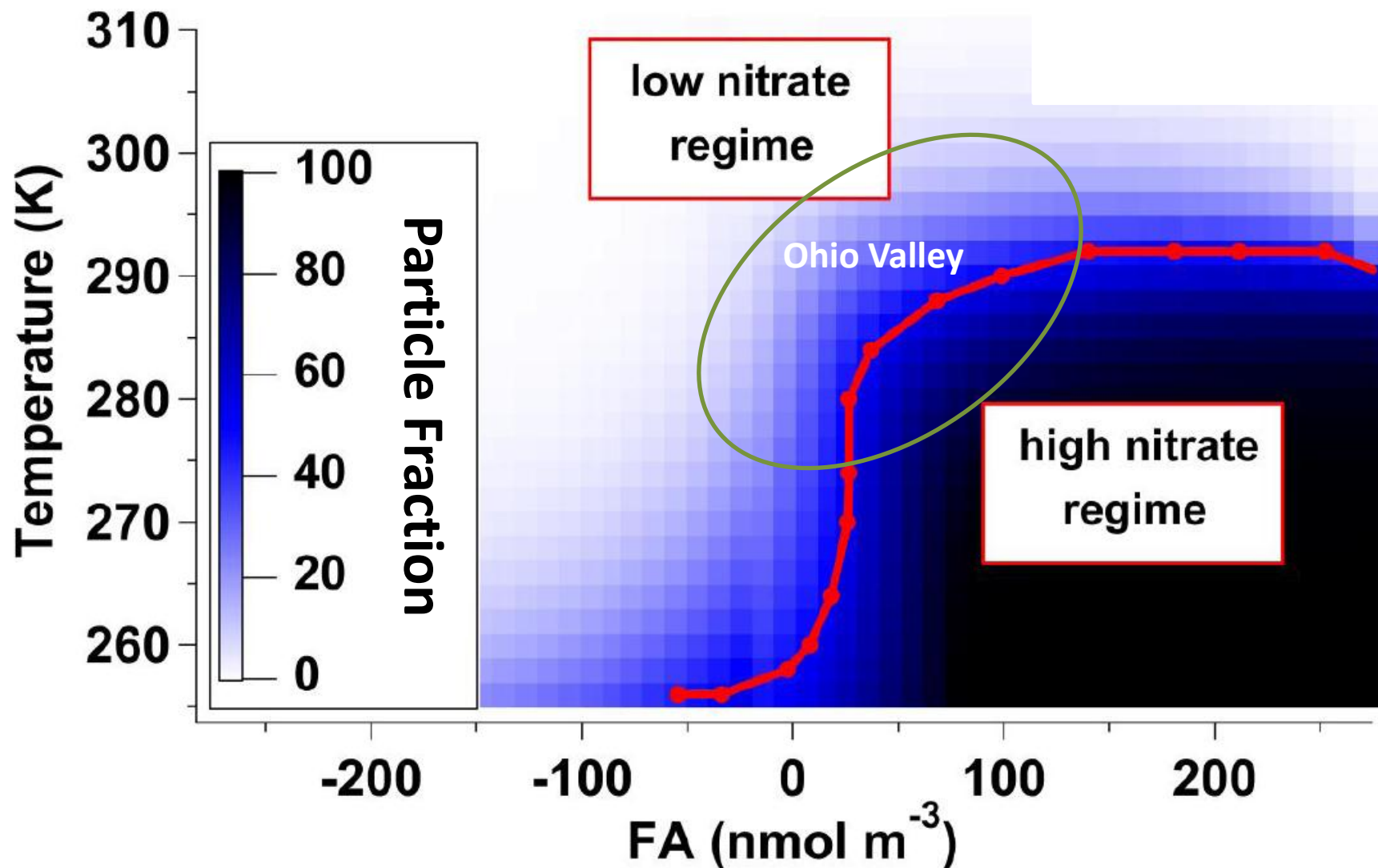


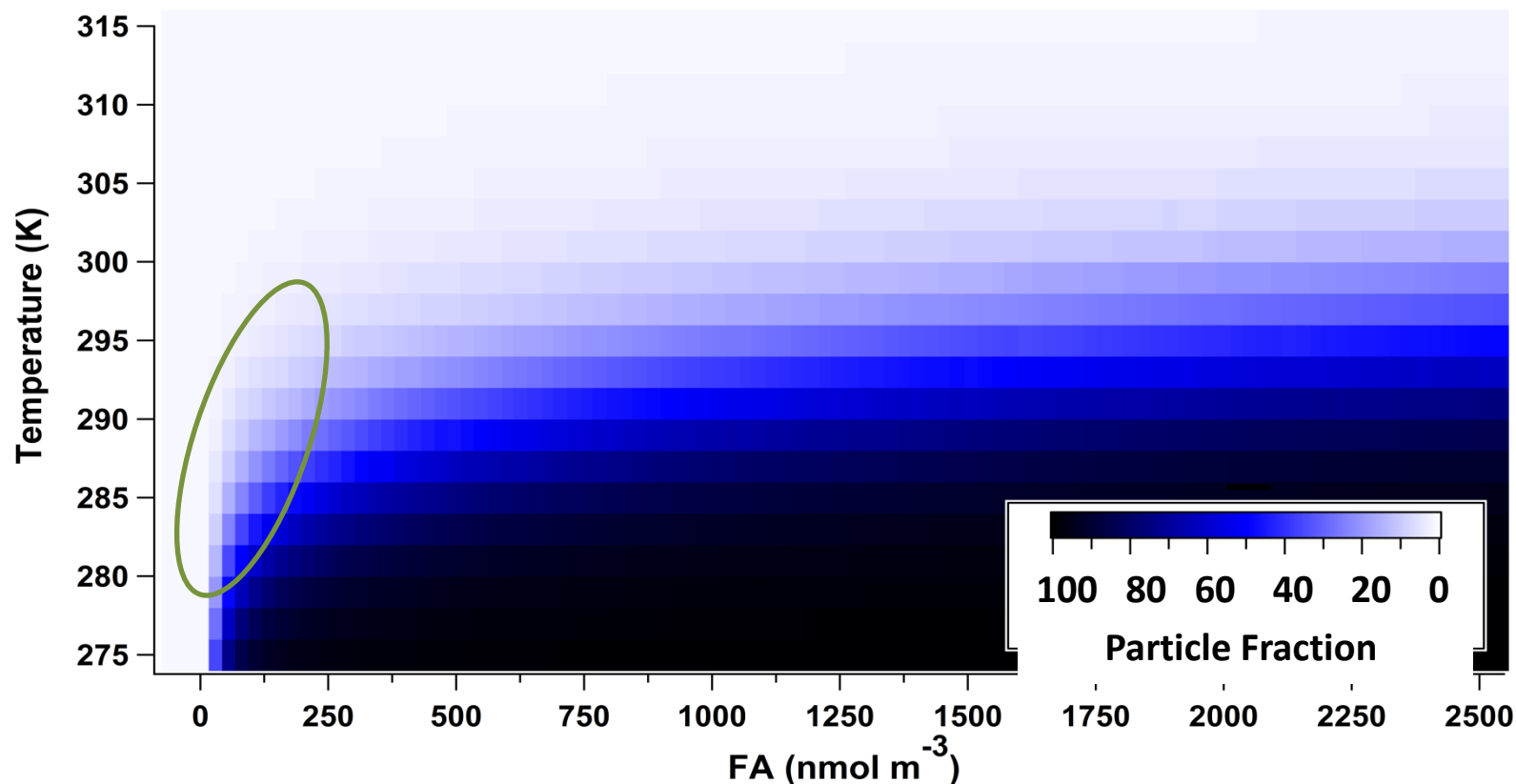
Figure 4-7. Seasonal variation of total ammonium ($\text{NH}_3 + \text{NH}_4^+$) concentration and $\text{NH}_3/\text{NH}_4^+$ ratio at Fresno during CRPAQS. Note that the y-axis on the right has a logarithm scale.

$$\text{FREE AMMONIA, } \text{FA} = \text{TA} - 2 * \text{TS} \\ = [\text{NH}_4^+] + [\text{NH}_{3(\text{g})}] - 2 * [\text{SO}_4^{2-}]$$



High NH_3 , low TS and TN

Nitrate partitioning driven mainly by meteorology



Conditions:

**RH = 40 %, T = 275 – 315 K, TN = 20 nmol m^{-3} , TS = 21 nmol m^{-3} , TA = 0 - 2750 nmol m^{-3}
FA values during the campaign = (-42 nmol m^{-3} to 2750 nmol m^{-3})**

Pasadena data on chemical map generated assuming $TN = 100 \text{ nmol m}^{-3}$
TN actually ranges from 10 – 700 and correlates with FA

